Introduction

This game is a revision activity to the content of primary chemistry and materials topics.

Running the activity

There are 50 cards, two to a page, all different. Print out as many pages as you need and cut them in half to make individual cards. Give out individual cards to each pupil. The cards can be laminated and a wax pencil used to mark them.

The teacher has the sheet of key word definitions. Mark or tick off the questions asked during each session. You may wish to substitute definitions targeted at your pupils. The definitions are read out and pupils have to recognise and cross off the key word on their card. The first pupil to cross off all the words on their card receives a small prize. Check the winning card with the class to focus on the words used in the activity. Pupils can write out any definitions they definitions they do not recognise.

For a blank file contact nigel.heslop@scienceyear.com

Safety

Not applicable.

More ideas

The questions can be used as the basis of a quiz. Key words could be displayed beside the teaching station. Sticky Velcro patches make a good support for the word display. There should only be a few key words to focus attention on the target vocabulary for that session.

Learning outcomes

Review pupil knowledge of:

- Changes of state
- Dissolvina
- Solvents
- Gases
- Irreversible changes

Where the activity fits in

Revising and consolidating. QCA SoW 5C, 5D, 6C and 6D.

Skills

Vocabulary, recall skills.

✓ Tick these off when used in the session.

The solid that dissolves in a liquid:

The liquid that does the dissolving:

Solvent

The mixture of dissolved solid and liquid:

A material that has a fixed shape:

Solid

A material that has a fixed volume but

not a fixed shape: Liquid
A material that does not have a fixed volume or shape: Gas

When a solid disappears into a liquid:

Two or more different materials together:

Mixture

To make jelly dissolve more quickly you

make the water.. Hotter

Method used to separate small pieces of solid from

a liquid: Filtration

When a liquid become a gas: Evaporation

When a gas becomes a liquid: Condensation

The temperature that a liquid becomes a gas: Boiling point

When a solid becomes a liquid:

Freezing point of water:

0°C

Boiling point of water:

100°C

To make jelly dissolve more quickly you make

it into ...: Smaller pieces

When salt water boils away the salt is ...: Left behind

A material that contains only one type of particle:

Pure

When pancake mix is heated it goes solid.

This change is ...: Irreversible

Solute	Solvent	Solution		
Gas		Mixture		Filtration
Evaporation	Condensation		Melting	
100°C			Pure	Irreversible

Solute	Solvent		Solid	
Gas			Hotter	Filtration
Evaporation	Condensation			0° <i>C</i>
	Smaller pieces	Left behind	Pure	

Solute	Solvent			Liquid
	Dissolving	Mixture	Hotter	
Evaporation		Boiling point	Melting	
	Smaller pieces	Left behind		Irreversible

Solute		Solution	Solid	
	Dissolving	Mixture		Filtration
Evaporation		Boiling point		0° <i>C</i>
	Smaller pieces		Pure	Irreversible

Solute		Solution		Liquid
	Dissolving		Hotter	Filtration
Evaporation			Melting	0° <i>C</i>
		Left behind	Pure	Irreversible

Solute			Solid	Liquid
		Mixture	Hotter	Filtration
	Condensation	Boiling point	Melting	
100°C	Smaller pieces	Left behind		

	Solvent	Solution	Solid	
Gas	Dissolving	Mixture		
	Condensation	Boiling point		0° <i>C</i>
100°C	Smaller pieces		Pure	

	Solvent	Solution		Liquid
Gas	Dissolving		Hotter	
	Condensation		Melting	0°C
100°C	Smaller pieces			Irreversible

	Solvent		Solid	Liquid
Gas	Dissolving			Filtration
		Boiling point	Melting	0° <i>C</i>
100°C		Left behind	Pure	

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